Quantification of Pitolisant: A RP - HPLC study

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Abstract

Objective

An attempt was made to develop a simple, isocratic and robust reverse phase high performance liquid chromatographic method for Pitolisant.

Material and methods

An Inertsil octa decyl silane column and photo diode array detector were employed for the quantification of Pitolisant at 268 nm using a mobile phase consisting of acetonitrile and 0.1 % v/v formic acid (90:10 % v/v) at a flow rate of 1.0 mL/min. for the efficient elution of Pitolisant.

Results

The developed method was validated for various parameters and the linearity range was observed at 25-150 μ g/mL. The correlation coefficient, limit of detection and limit of quantification calculated from the calibration curve were 0.9998, 0.3 μ g/mL and 1.0 μ g/mL respectively. The mean percentage recovery was 99.9 %. The formulation was subjected to stress at various conditions resulting in minor degradants at different retention times.

Conclusion

An isocratic, fast, robust reverse phase – high performance liquid chromatographic method was developed which can be exploited for the estimation of Pitolisant in bulk and tablets.

Key words: Pitolisant, High performance liquid chromatography, Photo diode array detector, Limit of detection, Degradation studies

1. Introduction

Pitolisant (chemically 1- [3 - [3 - (4 - chloro phenyl) propoxy] propyl] piperidine; hydrochloride) ^[1] (fig. 1) is an antagonist/ inverse agonist of the human histamine H3 receptor, used for the treatment of narcolepsy with or without cataplexy in adults ^[2]. Pitolisant is soluble in alcohol, acetonitrile and has a dissociation constant of 9.67.

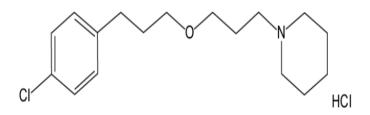


Figure 1. Chemical structure of Pitolisant

Literature review revealed that there is only one reverse phase – high performance liquid chromatography (RP - HPLC)^[3] method was reported using waters X- bridge phenyl (150 mm × 4.6mm, 3.5 μ m) column with photo diode array detector comprising of 0.1% of ortho phosphoric acid and acetonitrile (30:70 % v/v) as mobile phase at 1.0 mL/min. flow rate at 210 nm wavelength detection with 4.358 min. retention time and a single liquid chromatography – mass spectrometry/ mass spectrometry (LC-MS/MS)^[4] method reported for Pitolisant. So, an economically newer method development was undertaken in this study.

2. Materials and methods

Chemicals and reagents

The chemicals and reagents formic acid (analytical grade) and acetonitrile (HPLC grade) utilized for the study were purchased from local market. Pitolisant standard was obtained from Glenmark pharmaceuticals private limited, Mumbai and sample tablets were obtained from Zydus Cadila health care limited, Secunderabad.

Instrumentation

The proposed work was executed on a waters autosampler high performance liquid chromatography (HPLC) equipped with Inertsil octa decyl silane C18 column (250×4.6 mm, 5 µm), photo diode array (PDA) detector operated with empower software. Shimadzu balance and smart ultrasonic bath were also used.

Preparation of 0.1 % v/v formic acid

1.0 mL formic acid was pipetted to a 1000 mL volumetric flask, dissolved and made up with HPLC water.

Preparation of stock solution

25 mg of standard was weighed, transferred to a clean 25 mL volumetric flask. To this, a few mL of acetonitrile (ACN) was added, dissolved and made up with ACN (1000 μ g/mL). All

further dilutions of working standard (100 μ g/mL) and 25-150 μ g/mL were prepared from this stock solution.

Preparation of sample solution

20 Pitolisant (Wakix -18 mg) film coated tablets were weighed, powdered finely using mortar and pestle and a 25 mg equivalent was taken into a 25 mL clean volumetric flask. To this a few mL of ACN was added, sonicated for 1 hour, made up with ACN and filtered using 0.45 micron filter. This solution was further diluted as required.

Method Development

Chromatographic method conditions

Few trials were executed to develop a simple, economic HPLC method for Pitolisant using various columns, flow rates, diluent ratios etc. A few trials were conducted with Waters X – bridge C8 column ($150 \times 4.6 \text{ mm}$, 3.5μ) using ACN: 0.1% v/v tri - fluoro acetic acid (TFA) (80:20 % v/v), Agilent C18 column ($150 \times 4.6 \text{ mm}$, 3.5μ) and Inertsil C18 column ($250 \times 4.6 \text{ mm}$, 3.5μ) with ACN:0.1 % v/v TFA (70:30 % v/v) as mobile phase at a flow rate of 1 mL/min. Symmetrical peak shape, USP plate count and USP tailing were not satisfactory during the above trials. Then finally the method was optimized and developed using the Acetonitrile and 0.1 % v/v formic acid (90:10 % v/v) mixture at a flow rate of 1.0 mL/min. operated over a run time of 6 minutes and with an injection volume of $10 \mu \text{g/mL}$.

Method validation

The method developed was validated^[5] for parameters like system suitability, precision, linearity, limit of detection & limit of quantification, accuracy, specificity and robustness. The validation results obtained were compared with the acceptance criteria as per international council for harmonization (ICH) Q2 R1 guidelines.

System suitability

The suitability of the developed method and the HPLC system were tested with 100 μ g/mL prepared from standard stock (1000 μ g/mL).

Linearity

From the standard stock (1000 μ g/mL), volumes of 0.25, 0.50,0.75, 1.0,1.25, 1.50 mL each were pipetted into 6 labeled 10 mL volumetric flasks, dissolved and made up with diluent. Each solution was poured into labeled vials, placed in HPLC vial tray and analyzed.

Limit of detection (LOD) & limit of quantification (LOQ)

The limit of detection and limit of quantification were calculated by S/N ratio.

Precision

Six different solutions of 100 μ g/mL were prepared and analyzed. The peak areas obtained from the chromatograms during the same day and different days were noted and the % RSD was calculated.

Accuracy

The % recovery value was calculated at 3 levels (50 %, 100 %, 150 %) by standard addition method. To 1 mL sample stock taken into 10 mL volumetric flasks, 0.5 mL, 1.0 mL and 1.5 mL of 100 μ g/mL standard was added at each level respectively (3 concentrations at each level)and made up with diluent. The solutions were transferred to clean labeled vials placed in a HPLC tray and analysed.

Specificity

The interference of the excipients, solvents, degradants and impurities in the formulation were observed by comparing the chromatograms obtained by injecting a blank, a standard $(100 \,\mu\text{g/mL})$ and a placebo.

Robustness

Any change in chromatographic parameters such as peak area, retention time, tailing factor and theoretical plates were observed by deliberate modifications of flow rate (\pm 5 mL/min.), mobile phase composition (\pm 5 % v/v) over the software command. A 1.0 mL standard stock (1000 µg/mL) was pipetted into a clean 10 mL volumetric flask, made up with diluents and analysed at the altered conditions.

Assay

The developed method was applied for calculating the amount of Pitolisant in tablet and compared with the labeled amount mentioned on the strip label. The sample solution as discussed above was further diluted and analyzed.

Forced degradation study

The sample solution was subjected to stress conditions like acid, base, oxidation, hydrolysis, heat, photolysis^[7]. The peak areas obtained were analysed and observed for any possible degradations. The efficacy of the method was also studied in the presence of degradants.

Acid degradation

A 1.0 mL of 1000 μ g/mL sample solution and 1.0 mL 1N HCl were pipetted, transferred into a 10 mL volumetric flask, heated at 60 0 C for 30 min., cooled and made up with diluent after neutralization with 1 mL 1N NaOH.

Basic degradation

To a clean 10 mL volumetric flask, 1.0 mL (1000 μ g/mL) of sample solution and 1.0mL of 1N NaOH was added, heated for 30 min. at 60 ^oC, cooled, neutralized with 1 mL 1N HCl and made up with diluent.

Hydrolytic degradation

An aliquot (1.0 mL) of sample solution was pipetted and transferred into a clean 10 mL volumetric flask, 1.0 mL of HPLC water was added, heated at 60 ^oC for 30 min. cooled and made up to 10 mL with acetonitrile.

Thermal degradation

To a clean 10 mL volumetric flask, 1.0 mL of sample solution was added, heated at $105 \, {}^{0}\text{C}$ for 30 min. and made up with diluent after cooling.

Photolytic degradation

In a clean 10 mL volumetric flask, 1.0 mL sample stock was taken, kept in UV chamber for 30 min. and made up with diluent.

Oxidation degradation

1.0 mL of sample stock was taken into 10 mL clean volumetric flask, 1.0 mL of 3 % v/v hydrogen peroxide was added, cooled after heating for 30 min. at 60 0 C and made up with diluents.

All the degradation solutions made up were poured into labeled vials, placed in vial tray and $20 \,\mu\text{L}$ of this solution was injected into the HPLC column at specified intervals of time.

Solution stability

The stability study ^[6] was conducted with a 100 μ g/mL sample solution at room temperature and 2-8 ⁰C.

3. Results and discussion

Method development

The chromatographic method was optimized after the trials with the below conditions. Various combinations of C8 and C18 columns employing a variety of mobile phases were used for the method development. Finally an optimized peak (fig. 2) satisfying all the system suitability parameters is obtained. The conditions are mentioned in table 1.

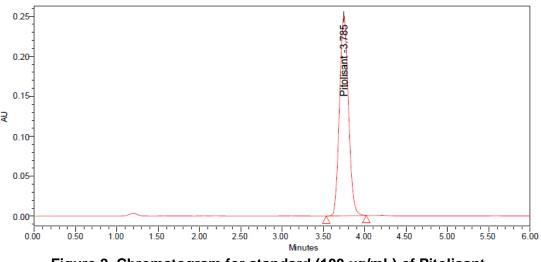


Figure 2. Chromatogram for standard (100 µg/mL) of Pitolisant

Parameters	Value
Column	Inertsil ODS (250 × 4.6mm, 5µm)
Mobile phase	ACN: 0.1 % v/v formic acid (90: 10 % v/v)
Mode of elution	Isocratic
Flow rate	1.0 mL /min.
Detection wavelength	268 nm
Volume of injection	10 µL
Run time	6 min.
Retention time	3.680 ± 0.45 min.

Table 1. Optimized chromatographic conditions of Pitolisant

ODS octa decyl silane, mm millimeter, μ m micrometer, ACN Acetonitrile, % v/v percentage volume by volume, mL/min. milliliter/minute, nm nanometer , μ L microliter, min. minute, \pm plus or minus

Method validation

The results obtained by validation complied with ICH specifications.

System Suitability

The parameters like theoretical plates, tailing, and % RSD were within the acceptance criteria as given in table 2. indicates the suitability of the method.

System suitability parameter	Acceptance criteria	Pitolisant
USP plate count	NLT 2000	4404
USP tailing	NMT 2.0	1.05
% RSD	NMT 2.0	0.79

Table 2. Results for System Suitability of Pitolisant

USP United states pharmacopoeia, NLT not less than, NMT not more than, % RSD percentage relative standard deviation

Linearity

The relationship between concentration and peak area was observed using the calibration curve and the range was found to be $25 - 150 \,\mu$ g/mL. The correlation coefficient, slope and y - intercept were calculated from the curve (fig. 3) and table 3.

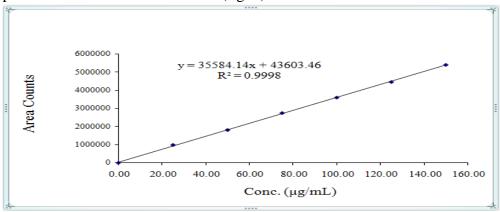


Figure 3. Linearity curve of Pitolisant

Conc.	Area
(µg/mL)	counts
25	985642
50	1810243
75	2745865
100	3596457
125	4452103
150	5396587

Table 3. Results for Linearity of Pitolisant

Conc. Concentration, µg/mL microgram/ milliliter

LOD & LOQ

The lowest amount detected and quantified was 0.3 μ g/mL, 1.0 μ g/mL respectively using S/N ratio.

Precision

The results obtained were precise as observed from the % relative standard deviation (0.79 for intraday and < 0.1 for inter - day study) which is in compliance with the ICH guidelines as in the table 4. (4A and 4B).

Conc.	Area	*Area	% Assay	*% Assay ± SD,
(µg/mL)	counts	counts	(%w/w)	% RSD
	3526451		99.6	
	3520130		99.8	
100	3535326	3539257	99.4	99.9 ± 0.78,
	3524856		99.5	0.79
	3596175		101.5	
	3532601		99.7	

Table 4A. Results for Precision (Intra – day) of Pitolisant

 μ g/mL microgram/milliliter, * Mean of six determinations, % percentage, % w/w percentage weight by weight, \pm plus or minus, SD standard deviation, % RSD percentage relative standard deviation

Table 4B. Results for precision (Inter-day)

Conc. (µg/mL)		Day 1	Day 2		Day 3	
	Area	*Assay (% w/w)	Area	*Assay (% w/w)	Area	*Assay (% w/w)
	counts	± SD,% RSD	counts	± SD,% RSD	counts	± SD,% RSD
Control	3530362	-	3538612	-	3523112	-
	3525101	99.8 ± 0.03,	3523451	99.6 ± 0.08,	3516351	99.9 ± 0.05,
	3524504	0.03	3520130	0.08	3519630	0.05
100	3525165		3525326		3517226	
	3524856		3524856		3518856	
	3527175	1	3527175		3520854	
	3525722]	3522601]	3520401	

Conc. Concentration, μ g/mL microgram/milliliter, * mean of 6 determinations, % w/w percentage weight by weight, \pm plus or minus, SD standard deviation, % RSD percentage relative standard deviation

Accuracy

The percentage recovery (100.4 % w/w) calculated was satisfactory as mentioned in the table 5.

%	Area	*Peak	Recovery	*Recovery (%) ± SD,
level	counts	area	(%)	%RSD
50	5292415		149.8	
	5282514	5272527	149.5	149.2 ± 0.79,
	5242653		148.3	0.79
100	7066421		199.9	
	7024715	7047733	198.8	199.4 ± 0.56,
	7052064		199.5	0.56
150	8841754		250.2	
	8842529	8848177	250.2	250.1 ± 0.17,
	8831247	1	249.9	0.17

Table 5. Results for Recovery (%) of Pitolisant

*Mean of three determinations, % percentage, ± plus or minus, SD Standard deviation, % RSD percentage relative standard deviation

Specificity

There were no interferences due to the excipients, solvents, degradants and impurities at the retention time of the analyte as observed from blank, placebo (fig. 4) below.

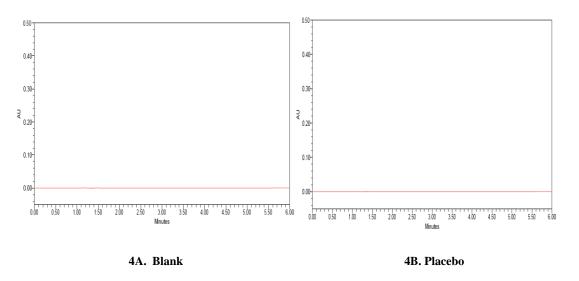


Figure 4. Chromatograms for Specificity of Pitolisant

Robustness

Deliberate modifications of flow rate and organic phase composition had little effect on USP plate count, USP tailing but much effect on the peak area as in the table 6. The method was found to be robust as the changes did not affect the assay and % RSD.

Parameter	Area	*Area	USP	USP	* % Assay ± SD,
	counts	counts	plate	tailing	% RSD
			count		
Flow rate	3784521		4361	1.06	100. 6 ± 0.2,
(+ 5 parts)	3768594	3776855	4325	1.08	0.2
	3777451		4396	1.08	
Flow rate	3475826		4456	1.11	99.9 ± 0.92,
(+ 5 parts)	3432658	3440900	4437	1.11	0.92
	3414215		4438	1.12	
Organic mobile phase	3865748		4312	1.09	100.1 ± 0.91,
composition	3895687	3861527	4339	1.10	0.91
(+ 5% v/v)	3823146		4313	1.09	
Organic mobile phase	3296587		4448	1.10	100.1 ± 0.67,
composition	3254581	3272707	4497	1.09	0.67
(- 5 % v/v)	3266953		4361	1.10	

Table 6. Results for Robustness of Pitolisant

* Mean of three determinations, USP United states pharmacopoeia, SD Standard Deviation, % RSD Percentage relative standard deviation

Assay

The percentage assay of the analyte in tablets was calculated and compared with the assay specified for Pitolisant by the manufacturer (table 7. and fig. 5).

Table 7. Results for Assay of Pitolisant

Sample	Label claim	Amount	% Assay (% w/w)
	(mg)	found (mg)	± SD
Wakix	18	18.1	100.4 ± 0.36

mg milligram, % percentage, % w/w percentage weight by weight, SD Standard deviation

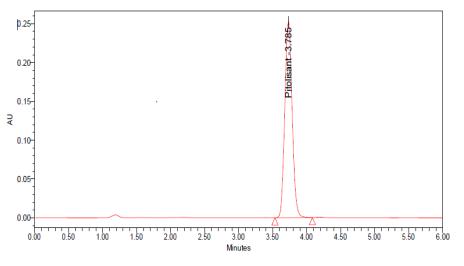


Figure 5. Chromatogram for Assay of Pitolisant

Solution stability

Stability study conducted showed negligible deviation up to 24 hr. both at room temperature and at 2-8 ^oC, the results are tabulated in table 8.

Condition	Stability a	t RT	Stability at 2 – 8 °C	
	% purity	% deviation	% purity	% deviation
0 hr.	100.0	0.00	99.8	0.20
6 hr.	99.5	0.50	99.2	0.80
12 hr.	99.4	0.60	98.9	1.10
18 hr.	99.6	0.40	98.2	1.80
24 hr.	98.0	2.00	97.6	2.40

Table 8. Results for Solution Stability

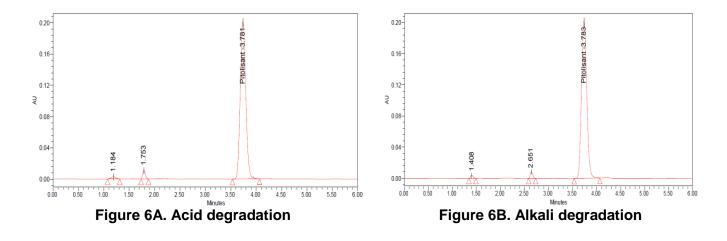
RT Room temperature, hr. hours, ^oC degrees Celsius, % percentage

Forced degradation studies

The degradation (%) at specified time intervals is provided in table 9. Major degradation occurred in acidic, alkali, oxidative conditions and minor degradation in thermal, photolytic hydrolytic conditions at 24 hr. as observed from the fig 6.

Table 9. Results for Forced Degradation Studies of Pitolisant

Type of degradation	Area counts	Degraded (%)	Purity angle	Purity threshold
*Control	3534168	0.2	-	-
Acid	2671047	24.4	0.264	10.542
Alkali	2699532	23.8	0.264	10.541
Peroxide	2622810	25.9	0.265	10.505
Hydrolysis	3421054	3.4	0.219	10.547
Photolysis	3386578	4.4	0.253	10.533
Thermal	3145642	11.2	0.217	10.154



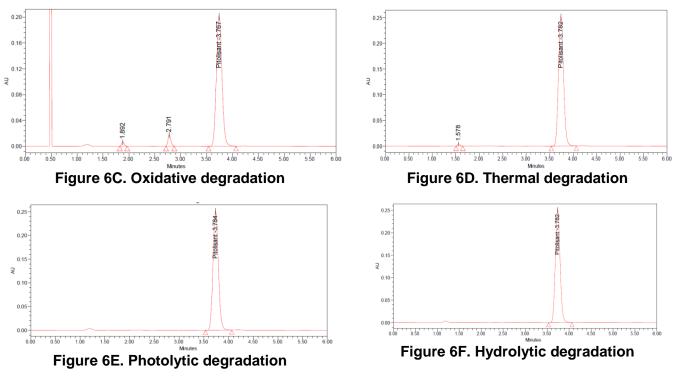


Figure 6. Chromatograms for Forced Degradation Studies

Conclusion

A simple, sensitive and economical isocratic RP-HPLC method was developed for quantification of Pitolisant and the results of validation were in accordance to ICH specifications. The proposed study can be considered as a rapid method as the analyte eluted at a less retention time than the previously reported method. The method was specific and stability indicating as few degradants were reported in all degradations. Thus, the present method can be routinely employed in the quality control testing of the analyte in active pharmaceutical ingredient and tablets.

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Ethics approval and consent to participate

Any kind of animals and humans were not used for the study

Availability of the data and material

The data and material will be available on request.

Contribution

The authors shared equal contribution in the preparation of manuscript and approved the manuscript.

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